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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/790,457	03/01/2004	Adam R. Pawloski	H1559	9956
45305 7590 05/31/2007 RENNER, OTTO, BOISSELLE & SKLAR, LLP (AMDS) 1621 EUCLID AVE - 19TH FLOOR			EXAMINER	
			SULLIVAN, CALEEN O	
CLEVELAND, OH 44115-2191		ART UNIT	PAPER NUMBER	
			1756	
			MAIL DATE	DELIVERY MODE
			05/31/2007	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

		Application No.	Applicant(s)				
Office Action Summary		10/790,457	PAWLOSKI ET AL.				
		Examiner	Art Unit				
		Caleen O. Sullivan	1756				
The MAILING DATE of this communication appears on the cover sheet with the correspondence address Period for Reply							
A SHO WHIC - Exter after - If NO - Failu Any r	ORTENED STATUTORY PERIOD FOR REPLY CHEVER IS LONGER, FROM THE MAILING DATE in the may be available under the provisions of 37 CFR 1.13 SIX (6) MONTHS from the mailing date of this communication. Period for reply is specified above, the maximum statutory period were to reply within the set or extended period for reply will, by statute, reply received by the Office later than three months after the mailing and patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION 16(a). In no event, however, may a reply be still apply and will expire SIX (6) MONTHS from cause the application to become ABANDO	ON. timely filed om the mailing date of this communication. NED (35 U.S.C. § 133).				
Status							
·	Responsive to communication(s) filed on <u>01 May 2007</u> .						
•	This action is FINAL . 2b) This action is non-final.						
3)	Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under <i>Ex parte Quayle</i> , 1935 C.D. 11, 453 O.G. 213.						
Disposition of Claims							
•	4)⊠ Claim(s) <u>1-23</u> is/are pending in the application.						
4a) Of the above claim(s) <u>7,14 and 22</u> is/are withdrawn from consideration.							
·	5) Claim(s) is/are allowed.						
· <u> </u>	☑ Claim(s) <u>1-6,8-13,15-21 and 23</u> is/are rejected. ☑ Claim(s) is/are objected to.						
	8) Claim(s) are subject to restriction and/or election requirement.						
Application Papers							
9) The specification is objected to by the Examiner.10) The drawing(s) filed on is/are: a) accepted or b) objected to by the Examiner.							
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).							
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).							
11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.							
Priority under 35 U.S.C. § 119							
12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).							
a) All b) Some * c) None of:							
 1. Certified copies of the priority documents have been received. 2. Certified copies of the priority documents have been received in Application No. 							
Certified copies of the priority documents have been received in Application No Copies of the certified copies of the priority documents have been received in this National Stage							
application from the International Bureau (PCT Rule 17.2(a)).							
* See the attached detailed Office action for a list of the certified copies not received.							
Attachment(s)							
1) Notice of References Cited (PTO-892) 4) Interview Summary (PTO-413)							
3) Infor	e of Draftsperson's Patent Drawing Review (PTO-948) mation Disclosure Statement(s) (PTO/SB/08) or No(s)/Mail Date	Paper No(s)/Mail 5) Notice of Informa 6) Other:	Date I Patent Application				

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DETAILED ACTION

Response to Amendment

1. Applicant's amendments to claims 1, 10, 16 and 21 as well as the addition of new claim 23 has not overcome the rejection under 35 USC 103(a), presented in the last Office Action. Therefore, Examiner restates the grounds of rejection presented in the last Office Action in response to the amendments.

Claim Rejections - 35 USC § 103

- The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 3. The factual inquiries set forth in *Graham* v. *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:
 - 1. Determining the scope and contents of the prior art.
 - 2. Ascertaining the differences between the prior art and the claims at issue.
 - 3. Resolving the level of ordinary skill in the pertinent art.
 - 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.
- 4. Claims 1-21 and 23 are rejected under 35 U.S.C. 103(a) as being unpatentable over Switkes et al in view of Wallace ('801) and further in view of Costantini ('317).

Switkes et al describes a study on the feasibility of immersion lithography at 157nm, which is a limitation recited in claims 5 and 10 and is within the range recited in claim 4. (See, abstract). In

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this study a class of commercially available liquids such as perfluoropolyethers are identified as good candidates as an immersion lithography medium, which meets the limitation of claims 2, 11 and 17. (See, abstract). Switkes et al goes on to disclose that the perfluoropolyethers are a good immersion lithography medium because they are transparent, optically clean, chemically inert and compatible with some current resist materials, which meets the limitations of claims 3, 12, and 16 where the immersion lithography medium is non reactive with the material on the surface of the semiconductor wafer and is substantially transparent to radiation. (See, Section II. pg. 2353; 2355).

Switkes et al also describes performing an immersion lithography process. Thin layers of resist are spun on a Si substrate and then baked. The substrate was then covered with a thin layer of immersion fluid and then exposed. Next a low molecular weight solvent was used to remove the immersion liquid. After removing the wafer was subjected to a post exposure bake, followed by a developing step in a TMAH solution. (See, Section III. Pg. 2355). This disclosure meets the limitations of claims 1, 10 and 16 where an immersion lithography medium is applied to the surface of semiconductor wafer and the material on the surface of the wafer is exposed to electromagnetic radiation, as well as the limitation of claims 9-10 and 16 where the material layer on the wafer is exposed through the immersion lithography medium. This disclosure also meets the limitation of claims 6, 13 and 18.

Switkes et al fails to disclose a step of applying supercritical CO₂ to the wafer to remove the immersion lithography medium from the surface of the wafer; however, Wallace discloses such a process step.

Wallace ('801) claims a method of processing a wafer comprising the steps of: placing the wafer have a wafer surface in an enclosed and controlled environment; exposing said wafer surface to a cleaning medium rendered as a supercritical fluid; purging said environment of substance

including soluble chemical compound liberated from said wafer surface by said cleaning medium. (See, claim 19). Wallace ('801) also discloses an example in which the supercritical fluid used is carbon dioxide. (See, col.8, 31-42). Wallace ('801) further discloses that removal of a material such as a fluorocarbon from the surface of a wafer could be facilitated by exposure to UV light during the exposure of the wafer to supercritical CO₂. These teachings in Wallace ('801) meet the limitation of claims 1, 10 and 16 where supercritical carbon dioxide is applied to a semiconductor wafer to remove immersion lithography medium from the surface of a semiconductor wafer.

Still, Switkes et al in view of Wallace ('801) fails to disclose the limitations of claims 1, 10 and 16, where after supercritical CO₂ is applied to the surface of the wafer a mixture of immersion lithography medium and supercritical CO₂ is removed from the surface, and the immersion lithography medium is recovered from the mixture and purified. Moreover, Switkes et al in view of Wallace ('801) fails to disclose the limitation of claims 8, 15 and 19 where the immersion lithography medium is recovered by reducing the temperature or pressure of the mixture to remove CO₂ from the mixture. Switkes et al in view of Wallace ('801) also fails to disclose that the recovered immersion lithography medium will exhibit the same chemical composition or the same purity as the lithography medium applied to the surface of the wafer as recited in claims 20-21 and 23, as well as the limitation of claims 10 and 23, where the purified immersion lithography medium is recycled. However, these limitations are taught in Costantini ('317).

Costantini ('317) discloses a supercritical fluid delivery and recovery system for semiconductor wafer processing. (See, col. 2, 7-11). In this method there is a recovery section (See, col. 3, 24-31) that takes in a solvent, which is a mixture of immersion fluid and supercritical CO₂, referred to as effluent, obtained from a semiconductor processing chamber. (See, col.6, 6-18). This disclosure teaches the limitations of claims 1, 10 and 16, where a mixture of the immersion

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lithography medium removed from the surface and the carbon dioxide is recovered. The recovery section functions to collect, separate and purify the by-product gas, the co-solvent and other contaminants in the effluent and then return them to their respective receiver tanks or discharge as waste, which meets the limitations of claims 1, 10 and 16 where the immersion lithography medium that is recovered and purified and the limitation of claims 10 and 23 where the immersion lithography medium is recycled. (See, col.4, 1-10).

Costantini ('317) further discloses that in the recovery section the effluent passes into a separator where pressure and temperature fall below the critical points and the effluent separates into a vapor phase and a liquid phase. (See, col.6, 21-25). This disclosure teaches the limitations of claims 8, 15 and 19. The vapor phase contains the gas or gas mixture originally supplied into the feed portion of the system. The liquid phase contains the solvent and any other suspended components still remaining, and it is passed into a separator and heated to its boiling point. Then the solvent is separated as a vapor back to a suitable purity to be reused in the semiconductor wafer process chamber. (See, col. 6, 29-56). Although, Costantini ('317) does not specify a purity for the recovered immersion fluid, it is inherent that "suitable for re-use" means the recovered fluid would exhibit the same chemical composition or purity as the immersion fluid applied to the surface of the wafer as recited in claims 20-21 and 23.

It would have been obvious to one of ordinary skill in the art at the time of invention by applicant to modify the teachings of Switkes et al in view of Wallace ('801) and further in view of Costantini ('317), in order to recover the immersion lithography medium that is removed by applying supercritical CO₂ to the wafer, because Wallace ('801) teaches that supercritical fluids such as supercritical CO₂ can be use to remove substances such fluorocarbons from the surface a semiconductor wafer, and Costantini ('317) teaches that one can recover and then purify effluent

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from a semiconductor wafer processing chamber in order to recycle the immersion lithography medium that is recovered back to the semiconductor wafer processing chamber for re-use, resulting in a more economically efficient semiconductor wafer processing method.

Response to Arguments

- 5. Applicant's arguments filed 05/01/2007 have been fully considered but they are not persuasive.
- 6. Applicant argues that the combination of Switkes et al, Wallace ('801) and Costantini ('317) fails to disclose the limitations of amended claims 1, 10, 16 and 21 as well as new claim 23 where the immersion lithography fluid recovered from the mixture is recovered, and purified to exhibit substantially the same purity as the original immersion lithography medium applied to the surface of the semiconductor wafer. Applicant further argues that the combination of references fails to also teach a step where the purified immersion lithography medium is recycled for use in the immersion lithography process. Lastly Applicant argues that the combination of references fails to disclose recovery of an immersion lithography fluid.

Although the combination of references, specifically Costantini ('317) doesn't explicitly state the effluent recovered from the semiconductor wafer processing chamber consists of immersion lithography fluid, one of ordinary skill in the art would appreciate that the combination of references, especially Costantini ('317), teaches or suggests that fluids and other materials, which includes immersion lithography fluids, can be recovered from a semiconductor wafer processing chamber, purified and then recycled for re-use in the process chamber. (See, col.6, 7-56). Costantini ('317) further states that the co-solvent recovered is separated into a vapor phase at a suitable purity to be reused in the process. One of ordinary skill in the art would appreciate that due to the sensitive nature of semiconductor wafer processing any fluid recycled to the process chamber would be of the same purity as when first introduced into the processing chamber.

Applicant has unduly limited the scope of the teachings of Switkes et al and Wallace ('801) and Costantini ('317). The combination of references is not only applicable for what is explicitly disclosed, but also for what one of ordinary skill in the art would understand is implicitly disclosed. Moreover, the rejection of claims 1-21 and 23 is an obviousness rejection over the combination of Switkes et al, Wallace and Costantini.

7. THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Caleen O. Sullivan whose telephone number is 571-272-6569. The examiner can normally be reached on 8:30am-5:00pm. If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Mark Huff can be reached on 571-272-1385. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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/COS/, 05/15/2007

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